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                 Substances (PICCS) has been added to CHEMLIST
                 New Extraction Code PAX now available in Derwent
NEWS
         Oct 27
                 Files
NEWS
        Oct 27
                 SET ABBREVIATIONS and SET PLURALS extended in
                 Derwent World Patents Index files
NEWS
        Oct 27
                 Patent Assignee Code Dictionary now available
                 in Derwent Patent Files
                 Plasdoc Key Serials Dictionary and Echoing added to
NEWS
        Oct 27
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                 Derwent announces further increase in updates for DWPI
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        Dec 15
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                 Merged CEABA-VTB for chemical engineering and
                 biotechnology
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        Dec 17
                 Corrosion Abstracts on STN
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        Dec 17
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         Dec 17
                 The CA Lexicon available in the CAPLUS and CA files
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=> s melamine/cns

2006 MELAMINE/CNS L1

=> file caplus

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SINCE FILE TOTAL

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SESSION ENTRY 4.26

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```
=> s l1/prep
          24368 L1
       2704895 PREP/RL
           2363 L1/PREP
L2
                   (L1 (L) PREP/RL)
=> s 12 and ammonia
         117416 AMMONIA
            144 L2 AND AMMONIA
1.3
=> s 13 and water
       1444929 WATER
             26 L3 AND WATER
=> d 14 1-26 bib abs
L4
     ANSWER 1 OF 26 CAPLUS COPYRIGHT 2001 ACS
AN
     2001:12430 CAPLUS
DN
     134:72330
TI
     Method for separating melamine from melamine-containing gas mixtures
IN
     Willems, Rob Gerard Jan
     DSM N.V., Neth.
PA
     PCT Int. Appl., 11 pp.
SO
     CODEN: PIXXD2
DT
     Patent
LA
     English
FAN.CNT 1
     PATENT NO.
                       KIND DATE
                                               APPLICATION NO. DATE
                                               -----
                                                                  20000524
     WO 2001000596
                                              WO 2000-NL354
                        A2 20010104
          W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR,
              CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU,
              LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD,
              SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU,
              ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ,
              CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
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PRAI NL 1999-1012467 19990629

AB The method for sepg. melamine from melamine-contg. gas mixts. obtained by synthesis of melamine from urea or thermal decompn. products of urea in the presence of ammonia and catalyst, comprises cooling the gas

mixt. in a cooling zone through direct contact with an evapg. medium contg. .gtoreq.1 water, ammonia, and ammonium carbamate, wherein the evapg. medium is sprayed so that specific area is >600 m2/m3 of liq., the gas has impulse (rhogas*(vgas)2) >0.2 kg/m-s2 in at least a part of the cooling zone, and the residence time of the gas in the cooling zone where the gas is cooled to a temp. below 210.degree., is <11 s.

```
L4 ANSWER 2 OF 26 CAPLUS COPYRIGHT 2001 ACS
```

AN 2000:855629 CAPLUS

DN 134:30951

TI Organic flame retardant compositions containing salts of nitrogen compounds with boron oxyacids

IN Blount, David H.

PA USA

SO U.S., 9 pp. CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE
US 6156240 A 20001205 US 1997-898931 19970723

PI US 6156240 A 20001205 US 1997-898931 19970723

AB An org. flame retardant material consists of: (1) a nitrogen-contg. salt of a boron oxyacid (contg. >1 nitrogen and >1 boron atoms), prepd. by reaction of 25-100 wt.% of one or more boron oxyacid or salt with 25-100 wt.% of a nitrogen-contg. salt-forming compd. and 10-25 wt.% water (optionally contg. aq. NH3), and (2) an addnl. carbonizable compd. selected from phosphorus and sulfur compds. that release acids upon pyrolysis. The compns. may also contain fillers (e.g., urea, melamine, amino phosphates, phenoplasts, sawdust, graphite, etc.). Nitrogen-contg. salt-forming compds., of component (1), are selected from ammonium carbonate, inorg. ammonium salts, amines, aminoplasts, thiourea, alkyl carbamates, sulfamic acids, nitriles, alkyl isocyanates, urea, amides,

and

polyamides. The compns. have use as flame retardants and as surfactants. RE.CNT 11 $\,$ RE

(1) Blount; US 5721281 1998 CAPLUS

- (2) Blount; US 5788915 1998 CAPLUS
- (3) Fox; US 5076969 1991 CAPLUS
- (4) Herndon; US 5151225 1992 CAPLUS
- (5) Kajander; US 5837621 1998 CAPLUS
- ALL CITATIONS AVAILABLE IN THE RE FORMAT
- L4 ANSWER 3 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 2000:475649 CAPLUS
- DN 133:91000
- TI Method and apparatus for separating component from gaseous medium by crystallization
- IN Paijens, Toine; Van Zee, Gerard; Vrijenhoef, Hans
- PA Kemira Agro Oy, Finland
- SO PCT Int. Appl., 16 pp. CODEN: PIXXD2
- DT Patent
- LA English

```
FAN.CNT 1
                   KIND DATE
                                        APPLICATION NO. DATE
    PATENT NO.
     -----
                                         ______
    WO 2000040566
                                        WO 1999-FI1089 19991229
PΙ
                    A1
                          20000713
        W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU,
            CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL,
            IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA,
            MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI,
            SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM,
            AZ, BY, KG, KZ, MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE,
            DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,
            CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                        FI 1998-2848
                     A 20000701
    FI 9802848
                                                          19981231
PRAI FI 1998-2848
                     19981231
    The method comprises injecting a gaseous medium contg. a component (e.g.,
    p-dichlorobenzene) to be sepd. into a cooling liq. (e.g., water)
    in a crystallizer to form free traveling vapor bubbles which upon cooling
    induce supersatn. of the crystq. component with subsequent crystn. of the
    component at the phase interface of the free traveling vapor bubbles in
    the cooling lig.
RE.CNT 4
RE
(1) Anon; Journal of Crystal Growth 1997, V178
(2) Henriette, W; Cryst Growth Org Mater 4, Int Workshop, 4th 1997, P206
(3) Kitamura, M; Crystal size control of sulfathiazole using high pressure
   carbon dioxide P378
(4) Saburo, K; US 4331826 A 1982 CAPLUS
L4
    ANSWER 4 OF 26 CAPLUS COPYRIGHT 2001 ACS
ΑN
    1999:597032 CAPLUS
DN
    131:230084
TI
    Acrylic polyorganosiloxane-based aqueous coating and its manufacture
IN
    Fukuji, Yoshihisa; Shikano, Miki
PA
    Toyo Ink Mfg. Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 10 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
FAN.CNT 1
    PATENT NO.
                    KIND DATE
                                        APPLICATION NO. DATE
     -----
                                         -----
                    A2 19990921
PI
    JP 11256071
                                        JP 1998-56306
                                                         19980309
AB
    The coating, having good water and oil resistance, comprises a
    copolymer of a monomer having an unsatd. double bond and a
    polyorganosiloxane group 5-95, a monomer having a double bond and an aq.
    sol. group 5-95 and other monomers 0-90% and a copolymer of 5-100% a
    monomer having a double bond and an aq. sol. group and other monomers
    0-95%. Thus, a coating was made from a mixt. of a copolymer, prepd. by
    polymn. of Silaplane FM 0721 45, Bu methacrylate (I) 30 and acrylic acid
     (II) 25% in 200 parts MEK in the presence AIBN, removing the solvent and
    adding an aq. soln. contg. formic acid and ammonia water
     , 1.44, a copolymer, prepd. by polymn. of I 20, MMA 60 and II 20 in
40:160
     (parts) iso-Pr alc. and MEK mixt. in the presence AIBN and adding an aq.
    soln. contg. formic acid and ammonia water, 57.2,
```

Denacol EX 810 5.6 and water 35.94%.

```
T.4
    ANSWER 5 OF 26 CAPLUS COPYRIGHT 2001 ACS
    1999:597031 CAPLUS
AN
DN
    131:230069
    Acrylic fluoropolymer aqueous coating and it manufacture
TI
    Fukuji, Yoshihisa; Shikano, Miki
IN
    Toyo Ink Mfg. Co., Ltd., Japan
PA
    Jpn. Kokai Tokkyo Koho, 10 pp.
SO
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
FAN.CNT 1
                    KIND DATE
    PATENT NO.
                                        APPLICATION NO. DATE
     ______
                                         _____
    JP 11256070
                    A2
                                         JP 1998-56305
                          19990921
PΙ
                                                         19980309
AB
    The coating, having good water and oil resistance, comprises a
    copolymer of a monomer having an unsatd. double bond and a fluoroalkyl
    group 5-95, a monomer having an unsatd. double bond and an aq. sol. group
    5-95 and other monomers 0-90% and a polymer of 5-100% a monomer having an
    unsatd. double bond and an aq. sol. group and 0-95% other monomers.
Thus,
    a coating was made from a mixt. of a copolymer, prepd. by polymn. of
    Viscoat 17FM 50, Bu methacrylate (I) 25 and acrylic acid (II) 25% in
    40:160 (parts) iso-Pr alc. and MEK mixt. in the presence AIBN, removing
    the solvent and adding an aq. soln. contg. formic acid and ammonia
    water, 1.44, a copolymer, prepd. by polymn. of I 20, MMA 60 and II
    20 in 40:160 (parts) iso-Pr alc. and MEK mixt. in the presence AIBN,
    removing the solvent and adding an aq. soln. contg. formic acid and
    ammonia water, 57.2, Denacol EX 810 5.6 and
    water 35.94%.
L4
    ANSWER 6 OF 26 CAPLUS COPYRIGHT 2001 ACS
    1999:421713 CAPLUS
AN
DN
    131:59512
ΤI
    Preparation of cyclic urea-formaldehyde polymer-modified
    phenol-formaldehyde and melamine-formaldehyde resin-based binders and
    Dupre, F. C.; Foucht, Millard E.; Freese, William P.; Gabrielson, Kurt
IN
D.;
    Gapud, Benjamin D.; Ingram, W. Hayes; McVay, Ted E.; Rediger, Richard A.;
    Shoemake, Kelly A.; Tutin, Kim K.; Wright, James T.
PA
    Georgia-Pacific Resins, Inc., USA
    PCT Int. Appl., 49 pp.
SO
    CODEN: PIXXD2
DT
    Patent
LA
    English
FAN.CNT 1
                   KIND DATE
    PATENT NO.
                                       APPLICATION NO. DATE
     -----
                                         -----
PΤ
    WO 9932534
                    A1 19990701
                                        WO 1998-US26922 19981218
        W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE,
            DK, EE, ES, FI, GB, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE,
```

KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

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RW: GH, GM, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES,
             FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI,
             CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                         US 1998-215742
                                                           19981217
                      Α
                           20000905
    US 6114491
                                          AU 1999-18313
    AU 9918313
                      A1
                           19990712
                                                           19981218
    BR 9813809
                            20001003
                                          BR 1998-13809
                                                           19981218
                      Α
     EP 1042382
                      A1
                           20001011
                                          EP 1998-963255
                                                           19981218
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, FI
PRAI US 1997-68286
                     19971219
                     19980804
     US 1998-95249
                     19981217
     US 1998-215742
     WO 1998-US26922 19981218
AB
     Phenol-formaldehyde and melamine-formaldehyde resins were modified with
     cyclic urea-formaldehyde prepolymers contg. .gtoreq.20% triazone and
     substituted triazone compds. and obtained from urea, formaldehyde and
     ammonia or primary amines by reacting into the base resin systems,
     blending with the completed base resin systems , or blending into the
base
     resin prepn. The modified phenol-formaldehyde and melamine-formaldehyde
     resins are useful as binders in variety of products including
     consolidation wood products (such as plywood), engineered lumber (such as
     laminated veneer lumber), insulators, laminates, abrasive coatings, etc.
     Thus, 1311 parts phenol was mixed with formaldehyde (50%) 583,
    water 1217, cyclic urea-formaldehyde prepolymer (prepd. from urea,
     formaldehyde and ammonia) 500, pearl starch 16, defoamer 1.5,
     and NaOH (50%) 158 parts to give resin having viscosity (25.degree.) 944
    cps, solids content 43.6%, refractive index 1.4643, Mn 279 and Mw 693,
    which was used as a adhesive in plywood.
RE.CNT 2
RE
(1) Nagoya Yuka KK; JP 07118355 A 1995 CAPLUS
(2) Nagoya Yuka KK; JP 08109309 A 1996 CAPLUS
    ANSWER 7 OF 26 CAPLUS COPYRIGHT 2001 ACS
T.4
AN
    1999:380601 CAPLUS
DN
    131:20497
TT
    Manufacture of water-resistant particleboards using urea
     resin-methylated formaldehyde-melamine-urea copolymer mixture adhesives
IN
    Kikuchi, Takeshi; Ito, Atsushi; Matsunaga, Koji
    Mitsui Chemicals Inc., Japan
PΑ
SO
    Jpn. Kokai Tokkyo Koho, 5 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
FAN.CNT 1
                   KIND DATE
                                          APPLICATION NO. DATE
    PATENT NO.
     -----
                 A2 19990615
PΙ
    JP 11156820
                                          JP 1997-322979 19971125
    Water-resistant particleboards are prepd. by mixing wood chips
AB
    with 8-3:2-7 (wt. ratio) blends of urea resin (I) and liq. methylated
    formaldehyde-melamine-urea copolymer (II) adhesive prepd. by polymg.
    2.2-6.6:1:0.1-1.0:0.05-1.0 (mol ratio) mixts. of HCHO, melamine, urea,
and
    MeOH and hot-pressing the mixts. A 7:1.0:1.13:0.5 (mol ratio) mixt. of
    HCHO, melamine, urea, and MeOH was polymd. at 80.degree. and subsequently
```

polymd. after adding 2.52:1.0:0.5:0.5 (mol ratio) mixt. of HCHO, melamine,

urea, and MeOH to the mixt. to give II. Chips of wood having the surface and core layers comprising lauan and an adhesive compn. contg. 3:7 I-II blend 100, NH4Cl 1, and 25% ammonia water 1 part were mixed and pressed 3.5 min at 185.degree. to give a particleboard exhibiting bending strength .apprx.21 N/mm2.

- ANSWER 8 OF 26 CAPLUS COPYRIGHT 2001 ACS L4
- 1998:712672 CAPLUS AN
- DN 130:14989
- Water-thinned resin dispersions and coatings with good stability and curability for beverage cans
- IN Yamaguchi, Kaoru
- Toyo Ink Mfg. Co., Ltd., Japan PΑ
- Jpn. Kokai Tokkyo Koho, 10 pp. SO CODEN: JKXXAF
- DTPatent
- Japanese LA
- FAN.CNT 1

KIND DATE DATE APPLICATION NO. DATE PATENT NO. _____ -----

- JP 1997-102855 A2 19981104 19970421 PΙ JP 10292027
- The title dispersions are manufd. by neutralizing acrylic-modified epoxy resins with amines and/or NH3 and then dispersing the neutralized resins in aq. solvents. The acrylic-modified epoxy resins are manufd. by (1) reacting (A) acrylic resins having monobasic carboxylic acid monomer

units with (B) arom. epoxy resins having epoxy groups and/or OH groups modified with alc. solvents, (2) polymg. (C) comonomers contg. ethylenically unsatd. double bond-having monobasic carboxylic acids in the presence of

(B), or (3) polymg. (C) with arom. epoxy resins having ethylenically unsatd. double bonds obtained by modifying (B) with (C). Coatings contg. the dispersions for beverage cans, are also claimed. Thus, styrene 68, Et

acrylate 13.5, and methacrylic acid 54 parts were polymd. in the presence of benzoyl peroxide in BuOH to 27.3%-solid acrylic resin soln. Sep., 400 parts Epikote 1007 was reacted with 200 parts ethylene glycol monobutyl ether and 200 parts BuOH in the presence of 0.05 part NaOH to obtain a modified epoxy resin soln., 304 parts of which was mixed with 139 parts the acrylic resin soln. and 10.3 parts dimethylethanolamine, reacted at 70.degree. for 2 h, mixed with H2O and 10 parts a 50%-solid bisphenol A-formaldehyde copolymer to give a 20.0%-solid dispersion showing good storage stability, mech. stability, and curability.

- ANSWER 9 OF 26 CAPLUS COPYRIGHT 2001 ACS L4
- AN 1998:535270 CAPLUS
- DN 129:218016
- Crosslinkable water-thinned coating compositions and manufacture TT
- Goto, Tokio; Amemoto, Masahide IN
- Dainippon Ink and Chemicals, Inc., Japan PA
- Jpn. Kokai Tokkyo Koho, 14 pp. SO
 - CODEN: JKXXAF
- DT Patent
- LA Japanese

```
FAN.CNT 1
                                       APPLICATION NO. DATE
                   KIND DATE
    PATENT NO.
     -----
                                        ______
    JP 10219193 A2 19980818 JP 1997-27615 19970212
PΙ
    The title compns. comprise (A) 100 parts (solids) emulsion-polymd. vinyl
AB
    polymers contg. OH and carboxy group and having Mn 15,000-35,000 and mol.
    wt. distribution .ltoreq.4; (B) 5-30 parts crosslinking agent(s) chosen
    from amino resins and polyisocyanates; and (C) 5-20 parts water
     -miscible org. solvents with water soly. .gtoreq.100 at
    20.degree. and b.p. 130-220.degree. under ambient pressure. A component
Α
    was prepd. from styrene 350, Et acrylate 430, Me methacrylate 100,
    2-hydroxyethyl methacrylate 100, and acrylic acid 20 parts, with
    solubilization by aq. ammonia and used with titania and Watersol
    S-695 and Cymel 303 to obtain a high-gloss coating with excellent impact,
    soiling, weather, water, acid, alkali, solvent, and corrosion
    resistance and adhesion.
L4
    ANSWER 10 OF 26 CAPLUS COPYRIGHT 2001 ACS
AN
    1998:197900 CAPLUS
DN
    128:231672
TI
    Polymer latexes for coating of automobile body covering sheets
IN
    Higuchi, Etsuo; Kawata, Rumi; Mori, Hidekazu
    Nippon Zeon Co., Ltd., Japan
PA
SO
    Jpn. Kokai Tokkyo Koho, 7 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
FAN.CNT 1
                                       APPLICATION NO. DATE
    PATENT NO.
                    KIND DATE
     JP 10081715 A2 19980331 JP 1996-255350 19960906
    JP 10081715
PΙ
    The latexes with excellent alkali-thickening property and compatibility
    with Al stearate powders are obtained by emulsion polymn. of monomer
    mixts. comprising (meth)acrylic acid esters 75-98, ethylenically unsatd.
    acids 1-4, (meth)acrylamides or their derivs. 1-3, and other comonomers
    0-23%. Coating compns. contg. the latexes and waterproofing agents are
    also claimed. Thus, a monomer mixt. comprising Et acrylate 62.5, Bu
    acrylate 30, acrylonitrile 2.5, acrylamide 1.2, N-methylolacrylamide 0.8,
    and acrylic acid 3 parts was polymd. in an aq. emulsion in the presence
of
    ammonium persulfate to give a copolymer latex, 100 parts (as solid) of
    which was blended with H2O 50, a polyethylene wax emulsion 10, melamine
    resin (M 3) 7, an amine catalyst (ACX) 0.9, a thickener 10, a 25% aq.
    ammonia 1.7, and Al stearate paste 34 parts to give a compn.
    showing viscosity 37,000 cP at pH 7.0 and its change .ltoreq.10% after 10
    days. A polyester fabric coated with the compn. showed soft handle, good
    abrasion resistance, and no surface tack.
L4
    ANSWER 11 OF 26 CAPLUS COPYRIGHT 2001 ACS
```

- AN 1998:126304 CAPLUS
- DN 128:168812
- TI Aqueous dispersion coating compositions for metal can interiors
- IN Nakamura, Tetsuhisa; Shiono, Teruo; Yamada, Masami; Harada, Sanji; Ihara, Masahiro; Tsuyama, Takeshi
- PA Toyo Ink Manufacturing Co., Ltd., Japan; Nakamura, Tetsuhisa; Shiono,

```
Teruo; Yamada, Masami; Harada, Sanji; Ihara, Masahiro; Tsuyama, Takeshi
    PCT Int. Appl., 28 pp.
SO
    CODEN: PIXXD2
DT
    Patent
LA
    Japanese
FAN.CNT 1
    PATENT NO.
                   KIND DATE
                                        APPLICATION NO. DATE
     -----
                                        -----
                                                        ______
                    A1
                          19980219
                                        WO 1996-JP2283
PΙ
    WO 9806782
                                                         19960812
        W: US
        RW: DE, FR, GB
                          19980729
                                        EP 1996-926631
    EP 855427
                    A1
                                                         19960812
        R: DE, FR, GB
                                        US 1998-51434
    US 6046256
                    Α
                          20000404
                                                         19980409
PRAI WO 1996-JP2283 19960812
    The title compns. comprise: (A) a modified epoxy resin as an emulsifying
    agent and (B) an arom. epoxy resin having Mn 1,000-30,000 as a resin to
be
    emulsified which is partially or completely neutralized with
    ammonia and/or amine, wherein the resin A is prepd. by reacting an
    arom. epoxy resin having an epoxy equiv. 2,000-20,000 with an acrylic
    resin comprising (meth)acrylic acid and having acid value 50-450. An
    acrylic soln. was prepd. from styrene 105, Et acrylate 105, methacrylic
    acid 90, benzoyl peroxide 3, BuOH 696.4 parts in the presence of benzoyl
    peroxide; a modified epoxy resin soln. from Epikote 1007 400, ethylene
    glycol monobutyl ether 300, and MEK 300 parts; an arom. epoxy resin soln.
    from PKHH 400, ethylene glycol monobutyl ether 300, and MEK 300 parts;
and
    a resol soln. from p-cresol 417.7, formalin (40% BuOH soln.) 580.1, and
    Mg(OH)2 2.2 parts with diln. by xylene-BuOH-cyclohexanone to 35% solids
    and used in 15:35:40:10 ratio in a coating compn.
L4
    ANSWER 12 OF 26 CAPLUS COPYRIGHT 2001 ACS
AN
    1997:543481 CAPLUS
DN
    127:149763
ΤI
    Modifying 1,3,5-triazine derivatives
    Tanaka, Norio; Fukue, Yasuo; Mizusawa, Kenichi; Ishikawa, Makoto
IN
    Nissan Chemical Industries, Ltd., Japan; Tanaka, Norio;; Fukue, Yasuo;;
PA
    Mizusawa, Kenichi;; Ishikawa, Makoto;
so
    PCT Int. Appl., 100 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    Japanese
FAN.CNT 1
    PATENT NO.
                   KIND DATE
                                       APPLICATION NO. DATE
     -----
                                        -----
   WO 9724338
                    A1 19970710
                                        WO 1996-JP3762 19961224
        W: AU, CA, CN, KR, NO, US
        RW: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT,
SE
                          19970710
    CA 2241525
                     AA
                                        CA 1996-2241525 19961224
                     A1
    AU 9711731
                          19970728
                                        AU 1997-11731
                                                         19961224
    AU 704558
                     B2
                          19990429
    EP 882720
                     A1
                         19981209
                                        EP 1996-942628
                                                         19961224
        R: AT, BE, CH, DE, ES, FR, GB, IT, LI, NL, SE
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19990127

CN 1996-199361

19961224

Α

CN 1206409

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JP 1996-346882
                                                               19961226
     JP 10231291
                       A2
                             19980902
                                            US 1998-91545
                                                               19980622
     US 6127538
                             20001003
                       Α
                             19980819
     NO 9802959
                                            NO 1998-2959
                                                               19980625
                       Α
PRAI JP 1995-340424
                       19951227
     JP 1996-340721
                       19961220
     WO 1996-JP3762
                       19961224
OS
     MARPAT 127:149763
     The title process involves reacting a s-triazine deriv. having .gtoreq.1
AB
     (monosubstituted) amino group on any of the ring carbon atoms with an
alc.
     by heating in the presence of a metallic catalyst and H to introduce an
     alkyl or alkenyl group into each (monosubstituted) amino group; reacting
а
     s-triazine deriv. having .gtoreq.1 (monosubstituted) amino group on any
of
     the ring carbon atoms with a dihydric alc. by heating in the presence of
     metallic catalyst and H to introduce a hydroxylated alkyl group into each
     (monosubstituted) amino group. Their unique phys. properties, e.g.,
soly.
     in water and in various org. solvents, high-temp. stability,
     m.p., b.p., and basicity are of great interest for wide-ranging
     applications, e.g., for modifier additives, in particular a
     flame-retardant and a plasticizer for resins. The reaction of 0.01 mol
     melamine with EtOH in the presence of 5% Pd/C under 10 kg/cm2 H at
     210.degree. for 2 h gave melamine conversion 11.0%, 9.1%
     2,4-diamino-6-ethylamino-s-triazine and 1.2%
2-amino-4,6-bis(ethylamino)-s-
     triazine.
L4
     ANSWER 13 OF 26 CAPLUS COPYRIGHT 2001 ACS
AN
     1997:394279 CAPLUS
DN
     127:6175
     Method for preparing water-dispersible acrylic resin
TТ
     Nishikawa, Katsue; Hagiwara, Youshichi
IN
PA
     W.R. Grace & Co. - Conn., USA; Nishikawa, Katsue; Hagiwara, Youshichi
     PCT Int. Appl., 32 pp.
SO
     CODEN: PIXXD2
DT
     Patent
LA
     English
FAN.CNT 1
                      KIND DATE
     PATENT NO.
                                            APPLICATION NO. DATE
PΙ
     WO 9714751
                      A1 19970424
                                            WO 1996-JP3037 19961021
         W: AL, AM, AU, AZ, BA, BB, BG, BR, BY, CA, CN, CU, CZ, EE, GE, HU,
             IL, IS, KE, KG, KR, KZ, LC, LK, LR, LS, LT, LV, MD, MG, MK, MN,
             MW, MX, NO, NZ, PL, RO, RU, SD, SG, SI, SK, TJ, TM, TR, TT, UA,
             \mathtt{UG},\ \mathtt{US},\ \mathtt{UZ},\ \mathtt{VN},\ \mathtt{AM},\ \mathtt{AZ},\ \mathtt{BY},\ \mathtt{KG},\ \mathtt{KZ},\ \mathtt{MD},\ \mathtt{RU},\ \mathtt{TJ},\ \mathtt{TM}
         RW: KE, LS, MW, SD, SZ, UG, AT, BE, CH, DE, DK, ES, FI, FR, GB, GR,
             IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML,
             MR, NE, SN, TD, TG
     JP 09124718
                      A2
                             19970513
                                             JP 1995-272368
                                                               19951020
     AU 9673340
                       Α1
                             19970507
                                             AU 1996-73340
                                                               19961021
PRAI JP 1995-272368
                       19951020
     WO 1996-JP3037
                       19961021
AB
     A water-dispersible resin exhibits excellent properties as a
```

base resin in aq. coating compn. while requiring a reduced amt. of org. solvents. The water-dispersible resin is prepd. by (1) reacting an epoxy resin having .gtoreq.2 1,2-epoxy groups/mol. and having an epoxy equiv. wt. 100-5000 with a (meth)acrylic acid polymer and a tertiary amine, while neutralizing the existing carboxyl groups before, during or after the reaction, to prep. an aq. dispersion of a modified epoxy resin having quaternary ammonium carboxylate groups, and (2) polymg., in this aq. dispersion, an unsatd. compd. having .gtoreq.1 unsatd. group/mol.

Epo

Tohto YD 017 was reacted with dimethylethanolamine, aq. Jurymer AC 10SH soln. was added to this modified epoxy resin to form quaternary ammonium carboxylate groups, the remaining carboxy groups of the acrylic resin

were

neutralized with aq. ammonia, and this aq. dispersant was used to polymerize Me methacrylate in the presence of initiator, forming coating binder.

- L4 ANSWER 14 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 1995:773256 CAPLUS
- DN 123:343550
- TI Water-thinned coating compositions with excellent corrosion and water resistance
- IN Shinohara, Chikaya; Ishida, Yoshinori; Takeda, Yasuyuki
- PA Toto Kasei Kk, Japan
- SO Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF
- DT Patent
- LA Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 07157711 A2 19950620 JP 1993-309305 19931209

The compns. comprise (A) 2-20 parts (based on 100 parts total compn.)

water-sol. resins obtained by copolymg. phosphoric ester-modified

epoxy resins and carboxy-contg. ethylenically unsatd. monomers, (B) aq.
thermosetting resins, and (C) hardeners. Thus, adding dropwise 804.5

parts di-Bu phosphate to a mixt. of 800 parts YD 128 and 200 parts YDPN
638 (phenol novolak epoxy resin) for 1 h, stirring them for 6 h, adding
95.6 parts neo acid and 0.1 part PPh3 to the mixt., and heating them at
150.degree. for 8 h gave a phosphoric ester-modified epoxy resin, 320
parts of which was dissolved in ethylene glycol mono-Bu ether (I),

treated

with 23.0 parts methacrylic acid and 57 parts Me methacrylate at 115-120.degree. in the presence of Bz2O2, dild. with I, and mixed with Et3N and H2O to give an aq. resin. A compn. of Almatex W 911 (acrylic resin) 33.3, Tipaque R 830 (TiO2) 54, Cymel 303 13, the aq. resin 10, and p-toluenesulfonic acid amine salt 0.3 part (solids) showed good storage stability and, applied on a Zn phosphate-treated steel plate and baked to form a coating, showed pencil hardness 4H and good water and salt-spray resistance.

- L4 ANSWER 15 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 1995:383028 CAPLUS
- DN 123:57580
- TI Fluidized-bed reactor process for manufacturing melamine from urea at elevated pressures

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Lee, Jing M.
IN
PA
     USA
SO
     U.S., 6 pp.
     CODEN: USXXAM
DT
     Patent
     English
LA
FAN.CNT 1
     PATENT NO.
                       KIND DATE
                                             APPLICATION NO. DATE
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                             -----
                                             US 1993-147848
     US 5384404
                       Α
                             19950124
                                                               19931105
PΙ
     WO 9620933
                       A1
                             19960711
                                             WO 1995-US36
                                                               19950103
         W: AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, ES, FI, GB, GE, HU, JP, KP, KR, KZ, LK, LU, MG, MN, MW, MX, NL, NO, NZ, PL, PT,
             RO, RU, SD, SE, SK, UA, US, UZ, VN
         RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG
                             19960724
                                             AU 1995-15972
                                                               19950103
     AU 9515972
                       A1
PRAI US 1993-147848
                       19931105
     WO 1995-US36
                       19950103
AB
     An improved process, summarized in 6 steps, for manufg. melamine from
urea
     simplifies the recovery of melamine, carbamate and ammonia from
     a fluidized bed reactor effluent stream by operating the process at 1.4-2
           In the process, a carbamate soln. can be produced at a sufficiently
     high concn. for use in a urea plant without an intervening concn. step.
     In addn., ammonia recycled as a fluidizing gas can be condensed
     against cooling water to permit easy sepn. of noncondensables
     such as oxygen used in the process as a passivator for carbamate
corrosion
     inhibition. The melamine product is produced as an aq. soln. free of
     melamine solids. Heat is recovered from the carbamate condensation and
     used for the vaporization of ammonia which is recycled to the
     reactor for fluidization.
     ANSWER 16 OF 26 CAPLUS COPYRIGHT 2001 ACS
T.4
AN
     1995:18473 CAPLUS
DN
     122:135906
     Properties of hydrophilic poly(ethylene terephthalate) modified by
TΙ
     grafting with poly(acrylic acid) followed by novel salts formation
     Konomi, Tsuyoshi; Shigaki, Mika; Sato, Yumi; Sakata, Keiko; Sugiura,
AU
     Hiroko
     Fac. Home Econ., Japan Women's Univ., Tokyo, 112, Japan
CS
     Sen'i Gakkaishi (1994), 50(3), 110-17
SO
     CODEN: SENGA5; ISSN: 0037-9875
     Journal
דת
     Japanese
LA
AB
     Novel salts of poly(ethylene terephthalate) (PET) modified by grafting
     with poly(acrylic acid) (PAA) were prepd. using salts of amines and Na
     salts of amino acid analog, as well as of alkali and alkali earth metal.
     Changes in hygroscopicity, tensile properties, bending rigidity of
     fabrics, electrostatic property, and thermal stability by the grafting
and
     the salt formation were investigated. In the case of amine and alc.
amine
```

salts, moisture regain were reduced due to the steric hindrance, in

adsorbing site for water mol., in the presence of bulky alkyl

groups attached to N atom. For the samples reacted with alkali metal salts, hygroscopicity increased with ionic strength of metal. By alkali metal salt formation, tensile strength decreased by 25% in comparison

with

that of PAA-grafted PET. Fibers reacted with novel salts retained the same tensile strength as that of PAA-grafted PET. PET fabrics modified showed slight decrease in bending rigidity. The activation energy for thermal degrdn. reaction was estd. The thermal stability of the above samples was generally reduced except the sample reacted with Ca ion.

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L4 ANSWER 17 OF 26 CAPLUS COPYRIGHT 2001 ACS
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AN 1992:7363 CAPLUS

DN 116:7363

TI Preparation of phenolic resins for laminating

IN Honda, Nobuyuki

PA Toshiba Chemical Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

1 JP 03162409 A2 19910712 JP 1989-302467 19891121

1 JP 06037541 B4 19940518

AB The title resins giving laminates with excellent flame, heat, and humidity

resistance, dimensional stability, punchability, and reduced warpage are prepd. by treating melamines with phenols and aldehydes in the presence of

NH3 or primary amines to prep. resols followed by treating the resols with

tertiary amines. Thus, melamine 150, PhOH 250, 37% formalin 350, and 40% aq. MeNH2 4 g were refluxed for 1 h to prep. a resol, which was refluxed with 3 g Et3N for 1 h, dehydrated under reduced pressure, dild. with MeOH-MePh mixt. and mixed with 341 g Ph3PO4 and 120 g tetrabromodiphenyl ether to prep. a varnish [solid resin content 60%, viscosity 2.0 P (25.degree.), gel time 4.5 min (150.degree.)]. The varnish was impregnated into 10 mils kraft paper to prep. prepregs (resin pickup 50%),

8 sheets of which were sandwiched between adhesive-backed Cu foils and heated at 170.degree. and 100 kg/cm2 for 75 min to obtain a 1.6-mm paper-based phenolic resin printed base showing solder resistance 50-60 s (JIS-C6481), water absorption 0.6% and surface temp. 50.degree. on punching, compared with 30-40 s, 0.9% and 60.degree. for a control similarly prepd. but omitting the refluxing with Et3N. The printed base also showed flame retardance (UL-94) V-0 and warpage 0.5-1.0 mm.

- L4 ANSWER 18 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 1990:632758 CAPLUS
- DN 113:232758
- TI Adducts of phosphonates and amines as fireproofing agents
- IN Von Bonin, Wulf; Von Gizycki, Ulrich
- PA Bayer A.-G., Fed. Rep. Ger.
- SO Ger. Offen., 12 pp.

CODEN: GWXXBX

DT Patent LA German

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI DE 3833977 A1 19900426 DE 1988-3833977 19881006

OS MARPAT 113:232758

- Adducts of phosphonates M[OPR(O)OR1]x (M = Group II-VIII metal; R = C1-6 aliph. group, C6-10 arom. group; R1 = H, C1-6 aliph. group, C6-10 arom. group; x = valence of M) and 0.05-1 mol amine/equiv OR1 group are prepd. and used as fireproofing agents for polymers, paper, wood, etc. Heating Al(OH)3 78, MeP(O)(OMe)2 373, and water 800 parts at 180.degree. gave Al[OPMe(O)OH]3 (I). A mixt. of 312 parts I (as 20% aq. soln.) and 126 parts melamine was heated 10 min at 100.degree., dried at 100-120.degree., and pulverized to prep. a fireproofing agent, which was mixed (200 parts) with 100 parts ethylene-vinyl acetate copolymer. The mixt. had limiting O index 45.5% and self-extinguishing time 3 s.
- L4 ANSWER 19 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 1989:463487 CAPLUS
- DN 111:63487
- TI Removal of phenol, melamine, and formaldehyde from wastewater
- IN Schwitzgelbel, Klaus
- PA USA
- SO U.S., 7 pp. CODEN: USXXAM
- DT Patent
- LA English

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI US 4824577 A 19890425 US 1987-86073 19870817

- AB A clean up process for removing melamine, HCHO, and phenol from wash water in plastics manuf. comprises pptg. reaction products from the phenol-HCHO wash stream at pH >9, contacting the filtrate with a C adsorbent bed to remove phenol, converting the HCHO to urotropine at pH >10 with excess NH3, and contacting the effluent with a C bed to remove urotropine. Similarly, the melamine-HCHO wash water is held at pH 4.5 and >50.degree. to ppt. reaction products and the resulting filtrate is treated as before to remove HCHO and residual melamine. The filtrates can be combined and treated together to form urotropine which can be removed by C adsorbents. Powd. C can act as a catalyst in the reaction. The treated effluent can be recycled to wash equipment.
- L4 ANSWER 20 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 1988:57091 CAPLUS
- DN 108:57091
- TI Base-catalyzed reactions of imino- or amino-functional compounds in preparation of polyether polyols
- IN Guettes, Bernd; Romanowski, Helmut; Mueller-Hagen, Gerd; Marquardt, Renate; Tischer, Gerlinde
- PA VEB Synthesewerk Schwarzheide, Ger. Dem. Rep.
- SO Ger. (East), 4 pp. CODEN: GEXXA8
- DT Patent
- LA German

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FAN.CNT 1
                                        APPLICATION NO. DATE
                    KIND DATE
     PATENT NO.
     -----
                                      DD 1982-245975 19821215
                     A1
    DD 242313
                           19870121
PΙ
    A precondensate prepd. from urea and/or a deriv., an aq. HCHO and
AB
     stabilized with NH3, an amine, or an amino alc. is treated, optionally in
     the presence of a catalyst, with an alkylene oxide, and the reaction
     product, after removal of water and, optionally, the addn. of a
     catalyst, is treated with addnl. alkylene oxide to prep. a polyether
    polyol which is useful in the prepn. of polyurethane foams. The method
     avoids cleavage and decompn. reactions during alkoxylation, and the
    precondensate used in the alkoxylation is homogeneous, pumpable, and
     stirrable. A mixt. of urea 300, 37% aq. HCHO 215, and aq. NH3 1 g was
     stirred 2 h to prep. a precondensate which (420 g) was treated with 6 g
     48% aq. KOH, heated to 110.degree., treated with 400 mL propylene oxide
     (I) during <1 h, distd. in vacuo to remove water, and treated
     with 600 mL I at 100-120.degree. to prep. a polyether polyol having OH
no.
     510, pH 9.8, viscosity (25.degree.) 3300 mPas, and water content
     0.1%.
    ANSWER 21 OF 26 CAPLUS COPYRIGHT 2001 ACS
AN
     1984:552877 CAPLUS
DN
    101:152877
ΤI
    Melamine-urea resins
    Nippon Kasei K. K., Iwaki, Japan
PA
SO
     Jpn. Kokai Tokkyo Koho, 8 pp.
     CODEN: JKXXAF
DT
     Patent
LA
    Japanese
FAN.CNT 1
                                         APPLICATION NO. DATE
     PATENT NO.
                    KIND DATE
     ---- ----
                           19840622 JP 1982-216396
PΙ
    JP 59108018 A2
                                                           19821210
    Melamine-urea resin [25036-13-9] varnishes, giving blocking-resistant
     decorative paper prepregs, are prepd. by heating between 85.degree. and
     the refluxing temp. a soln. contg. 1:3.0-5.0 mol ratio HCHO-urea in a
    water-MeOH [67-56-1] mixt. (which has been adjusted to pH 8.0-9.0
    with NH3 [7664-41-7] or NH3-hexamethylenetetramine [100-97-0] mixt.)
    until the pH of the soln. drops to 4.5-4.0, adjusting the pH of the soln.
     to 3.8-3.2 with HCOOH [64-18-6] and heating the soln. between 85.degree.
     and the refluxing temp. for 1.5-2.5 h, and adjusting the pH of the soln.
     to 8.5-9.5 and treating the soln. with melamine or a melamine-HCHO mixt.
    between 85.degree. and the refluxing temp. until the soln. pH drops to
     .apprx.7.5. The melamine-urea-methanol-HCHO mol ratio used was
     1.0:0.8-1.2:0.7-1.9:4.0-5.0;. Thus, a soln. (pH 8.0) contg. 37.8% HCHO 1118, MeOH 78, 25% aq. NH3 53, and urea 240 g was refluxed for 2 h until
     the pH of the soln. dropped to 4.2. After adding 3.7 g 20% aq. HCOOH,
the
```

soln. was refluxed for 2 h, mixed with 351 g 37.8% HCHO and 7.6 g 25% aq. Na2CO3 to adjust the soln. pH to 8.8, and treated with 554 g melamine at 85.degree. for 1.5 h. The polymer product was mixed with 1.5 g 25% aq. Na2CO3 and 23 g ethylene glycol to give a varnish (solids content 50.5%) having viscosity 46.0 cP at 25.degree. and pot life 15 days at 15-25.degree.. Paper impregnated with the resin and p-toluenesulfonic acid curing agent exhibited blocking temp. 35.degree. at 80% relative

humidity.

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L4 ANSWER 22 OF 26 CAPLUS COPYRIGHT 2001 ACS
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AN 1983:597421 CAPLUS

DN 99:197421

TI Separation of ammonia and carbon dioxide from mixtures containing ammonia, carbon dioxide and water

IN Goorden, Josephus Johannes Pentrus; Laurens, Jan Simon; Biermans, Andreas Johannes

PA Stamicarbon B. V., Neth.

SO Eur. Pat. Appl., 24 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

		_													
	PA	rent :	NO.		KI	ND	DATE			API	PLICAT	ION :	NO.	DATE	
ΡI	EΡ	8847	8		A:	1	1983	0914		EP	1983-	2003	09	19830	303
	ΕP	8847	8		B	1	1986	0611							
		R:	ΑT,	BE,	DE,	FR,	GB,	IT,	NL,	SE					
	NL	8200	905		Α		1983	1003		NL	1982-	905		19820	305
	AT	2033	5		E		1986	0615		AT	1983-	2003	09	19830	303
PRAI	NL	1982	-905		19	8203	305								
	מס	1002	200	200	10	2202	0.00								

EP 1983-200309 19830303

AB The NH3, CO2, and H2O in a mixt. such as obtained in the synthesis of melamine or urea are recovered sep. In a 1st zone NH3 free of CO2 and H2O

is sepd. by distn. The soln. remaining is transferred to a 2nd zone where

CO2 free of NH3 and H2O is sepd. by distn. The soln. remaining after CO2 recovery is fed to a 1st desorption zone where gaseous NH3, CO2, and H2O and a dil. soln. of NH3, CO2 and H2O are formed.. The latter dil. soln. is transferred to a 2nd desorption zone where the remaining CO2 and NH3 are removed and pure H2O is obtained. The NH3 and CO2 from the 2 desorption zones are recycled to the sepn. zones.

L4 ANSWER 23 OF 26 CAPLUS COPYRIGHT 2001 ACS

AN 1976:90939 CAPLUS

DN 84:90939

TI Treating waste mother liquor for production of melamine

IN Fujiyoshi, Kenji

PA Showa Denko K. K., Japan

SO Japan., 3 pp. CODEN: JAXXAD

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	JP 50026553	B4	19750901	JP 1967-44170	19670711 [*]
AB	Cyanuric acid (II) [10	8-80-5] and CO2	[124-38-9] were	added, preferably

high temp., to a waste mother liq. for prodn. of melamine (I) [108-78-1] by the urea method to crystallize out I (which adversely affected formation of urea) as melamine cyanurate (III) [37640-57-6] and to neutralize NH3 [7664-41-7] to form the carbonate. Neutralization of NH3

accelerated formation of III. Thus, 100 parts of aq. II soln. was added to 100 parts of waste mother liq. (NH3 12, CO2 8, urea 15, I 0.3, and water 64.7%) with stirring, followed by addn. of 16 parts CO2.

The pptd. product was removed by filtration, and I content of the filtrate

was 0.0075%, which corresponded to 95% removal of I.

- L4 ANSWER 24 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 1972:72986 CAPLUS
- DN 76:72986
- TI Separating melamine from a hot melamine vapor-containing synthesis gas mixture
- IN Verstegen, Johannes D. M.; Van Nassau, Petrus J. M.
- PA Stamicarbon N. V.
- SO Ger. Offen., 11 pp. CODEN: GWXXBX
- DT Patent
- LA German

FAN.CNT 1

PA	TENT NO.	KIND	DATE	API	PLICATION NO.	DATE
PI DE	2116200	 А	19711021	DE	1971-2116200	19710402
DE	2116200	C2	19820121			
NL	7004765	Α	19711005	NL	1970-4765	19700403
NL	169181	В	19820118			
NL	169181	С	19820616			
US	3711479	A	19730116	US	1971-130428	19710401
BE	765201	A1	19711004	ΒE	1971-101720	19710402
ZA	7102145	Α	19711229	ZA	1971-2145	19710402
FR	2089100	A5	19720107	FR	1971-11735	19710402
BR	7101994	A0	19730417	BR	1971-1994	19710402
AT	307432	В	19730525	AT	1971-2819	19710402
ES	389828	A1	19730601	ES	1971-389828	19710402
CA	944351	A1	19740326	CA	1971-109446	19710402
SE	381659	В	19751215	SE	1971-4355	19710402
NO	133233	В	19751222	NO	1971-1284	19710402
GB	1288697	A	19720913	GB	1971-1288697	19710419
PRAI NL	1970-4765	197004	103			

AB In a process for sepg. melamine [108-78-1] from a hot, synthesis gas mixt.

contg. melamine vapor, NH3, and CO2, the hot, melamine vapor-contg. gas was cooled by contacting it with an aq. soln. contg. 1.5-5 kg NH2CO2NH4 for every kg melamine to be recovered from the synthesis gas mixt. The melamine-free, NH3 and CO2-contg. gas mixt. contained less water (<20%) than similar gas mixts. obtained by conventional processes in which

the synthesis gas mixt. was cooled with water.

- L4 ANSWER 25 OF 26 CAPLUS COPYRIGHT 2001 ACS
- AN 1971:477711 CAPLUS
- DN 75:77711
- TI Regenerating waste gases from melamine production
- IN Kokubo, Ryo; Yokomichi, Koji; Takakuwa, Yasuo
- PA Nissan Chemical Industries, Ltd.
- SO Ger. Offen., 17 pp.

CODEN: GWXXBX

DT Patent LA German

FAN.CNT 1

					
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	DE 2053487	Α	19710527	DE 1970-2053487	19701030
	DE 2053487	B2	19741114		
	DE 2053487	C3	19750703		
	JP 50022020	B4	19750728	JP 1969-87379	19691101
	FR 2071884	A 5	19710917	FR 1970-39223	19701030
	GB 1309275	Α	19730307	GB 1970-51987	19701102

PRAI JP 1969-87379 19691101

AB The waste gases (mainly NH3 and CO2) generated during the conversion of urea to melamine at high temp. and pressure are used, directly or after the addn. of water or aq. (NH4)2CO3 soln. and at the pressure used for melamine prepn., for the prepn. of urea, and this urea synthesis soln. is freed of excess NH3 (which is recycled to the melamine synthesis)

and passed into the 1st decompn. stage of a urea synthesis process. No addnl. pump or compressor is needed. Thus, 29 kg urea/hr and 8.2 kg NH3/hr were heated at 400.degree. and 140 kg/cm2 to give 10 kg melamine/hr, and the waste gas contg. 16.4 kg NH3/hr and 10.6 kg CO2/hr

200.degree. was used for urea synthesis at 165.degree. and 135 kg/cm2, giving a mixt. of urea 7.2, NH3 12.3, CO2 5.3 and H2O 2.2 kg/hr, which was

freed of excess NH3 (4 kg/hr is recycled to melamine synthesis) and passed

into the 1st decompn. stage of a urea synthesis process.

L4 ANSWER 26 OF 26 CAPLUS COPYRIGHT 2001 ACS

AN 1970:12773 CAPLUS

DN 72:12773

at

TI Recycling of gases from melamine synthesis to urea synthesis in a combined

process for synthesis of melamine and urea

IN Kaasenbrood, Petrus J. C.; Van Nassau, Petrus J. M.

PA Stamicarbon N. V.

SO Ger., Offen., 10 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

I'M'. CIT I							
	PATENT NO.	KIND	DATE	AP	PLICATION NO.	DATE	
ΡI	DE 1904912	A	19690911	DE	1969-1904912	19690131	
	DE 1904912	B2	19770602				
	NL 6801577	Α	19690805	NL	1968-1577	19680202	
	BE 727645	Α	. 19690730	BE	1969-727645	19690130	
	AT 283379	В	19700810	AT	1969-980	19690131	
	FR 2001269	A5	19690926	FR	1969-2347	19690203	
	GB 1247951	Α	19710929	GB	1969-1247951	19690203	
	US 3682911	A	19720808	US	1969-796093	19690203	
PRAT	NL 1968-1577	19680	202				

AB A process for recycling the effluent from the synthesis of melamine (I) from urea is described. The I product gas, which contains I, NH3, and

CO2, is cooled directly with water or aq. solns. contg. NH3 and CO2, sepg. the I and giving a gaseous mixt. of NH3, CO2, and H2O. By cooling and adding, if necessary, H2O or aq. solns. contg. NH3 and CO2, solns. which contain NH4 carbamate or carbonate and free NH3 and are suitable for urea synthesis are obtained. The solns. are brought to urea-synthesis pressures, heated, and desorbed with a countercurrent CO2 gas stream. The NH3-CO2 gas mixt. from the desorption contained only small amts. of H2O vapor, and was fed the high-pressure step of urea synthesis. In this process, the amt. of H2O present in the NH3-CO2 mixts.

obtained from the I synthesis effluent is reduced.

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COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	62.48	66.74
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL SESSION
CA SUBSCRIBER PRICE	-15.29	-15.29

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PASSWORD:

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FULL ESTIMATED COST

SINCE FILE TOTAL ENTRY SESSION 0.15 0.15

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Structure search limits have been increased. See HELP SLIMIT for details.

=> s melamine/cn

L1 1 MELAMINE/CN

=> d l1

L1 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2001 ACS

RN 108-78-1 REGISTRY

CN 1,3,5-Triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN Melamine (8CI)

CN s-Triazine, 4,6-diamino-1,2-dihydro-2-imino- (6CI)

OTHER NAMES:

CN 1,3,5-Triazine-2,4,6(1H,3H,5H)-triimine

CN 2,4,6-Triamino-1,3,5-triazine

CN 2,4,6-Triamino-s-triazine

CN 2,4,6-Triaminotriazine

CN ADK Stab ZS 27

CN Cyanuramide

CN Cyanurotriamide

CN Cyanurotriamine

CN DG 002

CN DG 002 (amine)

CN Isomelamine

CN Mark ZS 27

CN Pluragard

CN Pluragard C 133

CN s-Triazinetriamine

CN Teoharn

CN Theoharn

CN Triamino-s-triazine

CN Triaminotriazine

CN Virset 656-4

CN Yukamelamine

CN ZS 27

FS 3D CONCORD

DR 504-18-7, 130392-03-9, 94977-27-2, 65544-34-5, 67757-43-1, 68379-55-5, 70371-19-6, 169314-62-9

MF C3 H6 N6

CI COM

LC STN Files: AGRICOLA, ANABSTR, APILIT, APILIT2, APIPAT, APIPAT2,
BEILSTEIN*, BIOBUSINESS, BIOSIS, BIOTECHNO, CA, CABA, CANCERLIT, CAOLD,
CAPLUS, CASREACT, CBNB, CEN, CHEMCATS, CHEMINFORMRX, CHEMLIST, CIN,
CSCHEM, CSNB, DDFU, DETHERM*, DIPPR*, DRUGU, EMBASE, GMELIN*, HODOC*,
HSDB*, IFICDB, IFIPAT, IFIUDB, MEDLINE, MRCK*, MSDS-OHS, NIOSHTIC,

PIRA,

PROMT, RTECS*, SPECINFO, TOXLINE, TOXLIT, TULSA, ULIDAT, USPATFULL, VTB (*File contains numerically searchable property data)
Other Sources: DSL**, EINECS**, TSCA**

(**Enter CHEMLIST File for up-to-date regulatory information)

4983 REFERENCES IN FILE CA (1967 TO DATE)
1749 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
4996 REFERENCES IN FILE CAPLUS (1967 TO DATE)
11 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

=> file caplus

COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 5.61 5.76

FULL ESTIMATED COST

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=> s 108-78-1 and cooling

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L3 5000 L2

217965 COOLING L4 109 L3 AND COOLING

=> d l4 1-109 bib

L4 ANSWER 1 OF 109 CAPLUS COPYRIGHT 2001 ACS

AN 2001:78371 CAPLUS

DN 134:132264

TI Production of melamine

IN Coufal, Gerhard

PA Agrolinz Melamin G.m.b.H., Austria

SO PCT Int. Appl., 13 pp.

CODEN: PIXXD2

DT Patent

LA German

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI WO 2001007421 A2 20010201 WO 2000-EP7093 20000725

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                CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
PRAI AT 1999-1299
                           19990727
      ANSWER 2 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
      2001:78370 CAPLUS
AN
DN
      134:132263
      Production of solid melamine
ΤI
      Coufal, Gerhard
IN
PA
      Agrolinz Melamin G.m.b.H., Austria
SO
      PCT Int. Appl., 12 pp.
      CODEN: PIXXD2
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LA
      German
FAN.CNT 1
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                                                   WO 2000-EP7092
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PRAI AT 1999-1298
                           19990727
L4
      ANSWER 3 OF 109 CAPLUS COPYRIGHT 2001 ACS
AN
      2001:12430 CAPLUS
DN
      134:72330
TI
      Method for separating melamine from melamine-containing gas mixtures
      Willems, Rob Gerard Jan
IN
PA
      DSM N.V., Neth.
      PCT Int. Appl., 11 pp.
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      NL 1012467
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PRAI NL 1999-1012467 19990629
    ANSWER 4 OF 109 CAPLUS COPYRIGHT 2001 ACS
    2000:842420 CAPLUS
AN
DN
    133:354102
    Activator for fly ash concrete
ΤI
    Yu, Youyi
IN
PA
    Peop. Rep. China
    Faming Zhuanli Shenqing Gongkai Shuomingshu, 7 pp.
SO
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    CN 1250757 A
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AN
    2000:705145 CAPLUS
DN
    133:269284
ΤI
    Melamine-containing corrosion inhibitor
IN
    Graichen, Stefan
PA
    Germany
    Eur. Pat. Appl., 7 pp.
SO
    CODEN: EPXXDW
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    German
LA
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                    A1 20001004
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    MARPAT 133:269284
RE.CNT 6
RE
(1) Ciba Geigy Ag; EP 0046139 A 1982 CAPLUS
(2) Ciba Geigy Ag; EP 0553962 A 1993 CAPLUS
(3) Darden, J; US 4647392 A 1987 CAPLUS
(4) McLaughlin, J; US 3976588 A 1976 CAPLUS
(6) Zts Chemie Gmbh; EP 0846690 A 1998 CAPLUS
ALL CITATIONS AVAILABLE IN THE RE FORMAT
    ANSWER 6 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
AN
    2000:475650 CAPLUS
DN
    133:89930
TI
    High-pressure process for preparation of melamine from urea
IN
    Vrijenhoef, Hans
PA
    Kemira Agro Oy, Finland
SO
    PCT Int. Appl., 18 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    English
FAN.CNT 1
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APPLICATION NO. DATE

PATENT NO.

KIND DATE

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WO 2000040567
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                          A 20000701 FI 1998-2847
                                                                      19981231
      FI 9802847
PRAI FI 1998-2847
                          19981231
RE.CNT 4
RE
(1) Dsm N V; WO 9734879 A1 1997 CAPLUS
(2) Kemira, O; WO 9501345 A1 1995 CAPLUS
(3) Roger, E; US 4565867 A 1986 CAPLUS
(4) Ryo, K; US 3484440 A 1969
L4
      ANSWER 7 OF 109 CAPLUS COPYRIGHT 2001 ACS
      2000:475649 CAPLUS
AN
      133:91000
DN
      Method and apparatus for separating component from gaseous medium by
TΙ
      crystallization
      Paijens, Toine; Van Zee, Gerard; Vrijenhoef, Hans
IN
      Kemira Agro Oy, Finland
PA
      PCT Int. Appl., 16 pp.
so
      CODEN: PIXXD2
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      FI 9802848
                          A 20000701
                                                FI 1998-2848
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PRAI FI 1998-2848
                          19981231
RE.CNT
(1) Anon; Journal of Crystal Growth 1997, V178
(2) Henriette, W; Cryst Growth Org Mater 4, Int Workshop, 4th 1997, P206
(3) Kitamura, M; Crystal size control of sulfathiazole using high pressure
     carbon dioxide P378
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      ANSWER 8 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
      2000:393779 CAPLUS
AN
DN
      133:5138
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Process for manufacture of melamine-formaldehyde resin for water soluble
      coatings
     Motiu, Iancu; Vaszilcsin, Ileana; Vuicin, Miriana; Matyasin, Marta;
IN
Lazar,
      Dorin; Serenciuc, Vasile
      S.C. "Azur" S.A., Timisoara, Rom.
PA
      Rom., 3 pp.
SO
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DT
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LA
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FAN.CNT 1
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PΙ
     RO 110510
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                                                                        19910624
     ANSWER 9 OF 109 CAPLUS COPYRIGHT 2001 ACS.
L4
      2000:351513 CAPLUS
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DN
      133:5397
ΤI
     Method for producing pure melamine
IN
     Coufal, Gerhard
     Agrolinz Melamin G.m.b.H., Austria
PA
      PCT Int. Appl., 18 pp.
      CODEN: PIXXD2
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               CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
PRAI AT 1998-1894
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RE.CNT
RE
(1) Agrolinz Melamin Gmbh; WO 9620182 A 1996 CAPLUS
(2) Agrolinz Melamin Gmbh; WO 9623778 A 1996 CAPLUS
(3) Agrolinz Melamin Gmbh; WO 9720826 A 1997 CAPLUS
T.4
     ANSWER 10 OF 109 CAPLUS COPYRIGHT 2001 ACS
AN
      2000:169504 CAPLUS
DN
      132:181422
ΤI
      Process and apparatus for manufacture of melamine
IN
      Jiang, Dazhou; Tan, Jianping; Jin, Yong; Yao, Wenhu; Dong, Lanzhong
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      Peop. Rep. China
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      Faming Zhuanli Shenqing Gongkai Shuomingshu, 9 pp.
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LA
FAN.CNT 1
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     ANSWER 11 OF 109 CAPLUS COPYRIGHT 2001 ACS
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DN
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IN
     Research Institute of Atomic Energy Utilization, Chinese Academy of
PA
     Sciences, Peop. Rep. China
     Faming Zhuanli Shenqing Gongkai Shuomingshu, 13 pp.
ŞO
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ΤI
     Method for preparing melamine
IN
     Wang, Yin
     DSM N.V., Neth.
PA
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RE.CNT 4
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(3) Melamine Chemicals Inc; EP 0747366 A 1996 CAPLUS
(4) Murata; US 3308123 A 1967
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L4
     1999:678213 CAPLUS
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     Scale inhibitor for cooling water or boiler water systems
ΤI
     Ando, Shinya; Ida, Yoshimi; Iimura, Akira; Ano, Shinji; Kibata, Kenji;
TN
     Usui, Urara
     Sanyo Chemical Industries, Ltd., Japan; Kurita Water Industries, Ltd.
PA
     Jpn. Kokai Tokkyo Koho, 6 pp.
SO
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     1999:595151 CAPLUS
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     High-pressure process for the manufacture of crystalline melamine
TI
IN
     Tjioe, Tjay Tjien
     DSM N.V., Neth.
PA
SO
     PCT Int. Appl., 28 pp.
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(2) DSM; EP 0808836 A 1997 CAPLUS
(3) Melamine Chemicals Inc; EP 0747366 A 1996 CAPLUS
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L4
AN
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TI
     Process for the preparation of melamine
ΑU
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     Res. Discl. (1999), 424 (Aug.), P1109-P1110 (No. 42474)
SO
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     131:145237
TI
     Method for cooling melamine
IN
     Coufal, Gerhard
PA
     Agrolinz Melamin G.m.b.H., Austria
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     PCT Int. Appl., 16 pp.
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(1) Kemira, O; WO 9501345 A 1995 CAPLUS
(2) Thomas, R; US 4565867 A 1986 CAPLUS
     ANSWER 17 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
AN
     1999:271343 CAPLUS
DN
     130:297367
TI
     Preparation of dry melamine powders by cooling melted melamines
     with liquid ammonia and their apparatus
IN
     Best, David Edward; Tjioe, Tjay Tjien
PA
     DSM N.V., Neth.
SO
     PCT Int. Appl., 17 pp.
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(1) Thomas, R; US 4565867 A 1986 CAPLUS
    ANSWER 18 OF 109 CAPLUS COPYRIGHT 2001 ACS
ΑN
    1999:116736 CAPLUS
DN
    130:125509
ΤI
    Process and equipment for manufacture of melamine
IN
    Jiang, Dazhou; Jin, Yong; Yu, Zhiqing; Yi, Jianglin
PA
    Peop. Rep. China
SO
    Faming Zhuanli Shenging Gongkai Shuomingshu, 8 pp.
    CODEN: CNXXEV
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    1999:113610 CAPLUS
AN
DN
    130:155746
ΤI
    A process for manufacture of stabilized alkali or alkaline earth metal
    hypobromite and its use in water treatment to control microbial fouling
IN
    Dallmier, Anthony W.; McCoy, William F.
    Nalco Chemical Company, USA
PA
SO
    PCT Int. Appl., 39 pp.
    CODEN: PIXXD2
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RE.CNT
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     ANSWER 20 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
AN
     1999:109783 CAPLUS
DN
     130:267044
ΤI
    Hydrothermal Synthesis of Organic Channel Structures: 1:1 Hydrogen-Bonded
     Adducts of Melamine with Cyanuric and Trithiocyanuric Acids
     Ranganathan, Anupama; Pedireddi, V. R.; Rao, C. N. R.
ΑU
     Chemistry Physics of Materials Unit, Jawaharlal Nehru Centre for Advanced
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     Scientific Research, Bangalore, 560 064, India
     J. Am. Chem. Soc. (1999), 121(8), 1752-1753
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    CODEN: JACSAT; ISSN: 0002-7863
     American Chemical Society
PΒ
DT
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RE.CNT
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(2) Mathias, J; J Am Chem Soc 1994, V116, P4316 CAPLUS
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(6) Wang, Y; J Crystallogr Spectrosc Res 1990, V20, P79 CAPLUS
ALL CITATIONS AVAILABLE IN THE RE FORMAT
     ANSWER 21 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
AN
     1999:81358 CAPLUS
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     130:113720
    Manufacture of conductive copper-nickel alloy powder, electric conductive
TT
    paste and electronic equipment
IN
    Nikaido, Ryuji; Ono, Taiichi; Nishiyama, Nobuyuki
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Alps Electric Co., Ltd., Japan; Teikoku Piston Ring Co., Ltd.
SO
     Jpn. Kokai Tokkyo Koho, 7 pp.
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AN
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ΤI
    Method for preparing melamine from urea with cooling using
     evaporating medium
    Tjioe, Tjay Tjien; Slangen, Hubertus Jozef Maria Dsm N. V., Neth.
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so
    PCT Int. Appl., 21 pp.
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(1) Agrolinz Melamin GMBH; WO 9620182 A 1996 CAPLUS
(2) Agrolinz Melamin GMBH; WO 9620183 A 1996 CAPLUS
(3) Agrolinz Melamin GMBH; WO 9623778 A 1996 CAPLUS
(4) Agrolinz Melamin GMBH; WO 9720826 A 1997 CAPLUS
(5) Best, D; US 5514796 A 1996 CAPLUS
ALL CITATIONS AVAILABLE IN THE RE FORMAT
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    ANSWER 23 OF 109 CAPLUS COPYRIGHT 2001 ACS
AN
    1998:806644 CAPLUS
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    130:53077
TI
    Crystalline melamine manufacture from urea in a high-pressure process
IN
    Tjioe, Tjay Tjien
PA
    Dsm N. V., Neth.
SO
    PCT Int. Appl., 30 pp.
    CODEN: PIXXD2
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    Agrolinz Melamin Gmbh; WO 9620183 A 1996 CAPLUS
    Agrolinz Melamin Gmbh; WO 9720826 A 1997 CAPLUS

(4) Best, D; US 5514796 A 1996 CAPLUS
(5) Dsm N V; EP 0808836 A 1997 CAPLUS
ALL CITATIONS AVAILABLE IN THE RE FORMAT
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     ANSWER 24 OF 109 CAPLUS COPYRIGHT 2001 ACS
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     1998:794994 CAPLUS
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     130:25724
ΤI
     Manufacture of melamine from urea via high-pressure process
IN
     Tjioe, Tjay Tjien; Slangen, Hubertus Jozef Maria
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     Dsm N.V., Neth.
     PCT Int. Appl., 20 pp.
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(5) Best, D; US 5514796 A 1996 CAPLUS
ALL CITATIONS AVAILABLE IN THE RE FORMAT
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L4
AN
    1998:789134 CAPLUS
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ΤI
    Method for preparing high-purity solid melamine with ammonia
    cooling stage
    Tjioe, Tjay Tjien
IN
    Dsm N.V., Neth.
PΑ
SO
    PCT Int. Appl., 17 pp.
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(3) Agrolinz Melamin GMBH; WO 9623778 A 1996 CAPLUS
(4) Agrolinz Melamin GMBH; WO 9720826 A 1997 CAPLUS
(5) Best, D; US 5514796 A 1996 CAPLUS
ALL CITATIONS AVAILABLE IN THE RE FORMAT
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ANSWER 26 OF 109 CAPLUS COPYRIGHT 2001 ACS
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     Process for reducing the amount of melam and melem in crude melamine
ΤI
     Jaeger, Emmerich; Zenkl, Ernst; Proemer, Sylvia; Kloeckl, Wolfgang
IN
     Agrolinz Melamin Gmbh, Austria
PA
SO
     Austrian, 4 pp.
     CODEN: AUXXAK
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     German
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TI
     Manufacture of melamine from urea
IN
     Van Wijck, Julius Gerardus Theodorus
     Dsm N.V., Neth.; Van Wijck, Julius Gerardus Theodorus
PA
SO
     PCT Int. Appl., 15 pp.
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     Intumescent thermoplastic polyamide graft polymers
     Moore, William Percy
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     Agrinutrients Company, Inc., USA
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     U.S., 8 pp.
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    1998:102852 CAPLUS
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DN
    128:128704
    Manufacture of highly pure solid melamine from urea melt by high-pressure
ΤI
    and cooling with ammonia
    Van Wijck, Julius Gerardus Theodorus
IN
    DSM N.V., Neth.; Van Wijck, Julius Gerardus Theodorus
PA
so
    PCT Int. Appl., 23 pp.
    CODEN: PIXXD2
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            RU, TJ, TM
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    Fibrous chemical compounds as intermediates of hexagonal boron nitride
and
    their manufacture
    Faustinus, Fauzi; Tani, Masato; Suzue, Masayoshi
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    Otsuka Chemical Co., Ltd., Japan
    Jpn. Kokai Tokkyo Koho, 6 pp.
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    128:62246
    Manufacture of highly pure melamine obtained from urea plant
TI
    Van Wijck, Julius Gerardus Theodorus
IN
    DSM N.V., Neth.; Van Wijck, Julius Gerardus Theodorus
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SO
    PCT Int. Appl., 23 pp.
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    High-pressure noncatalytic process for recovery of melamine from urea
ΤI
    pyrolysis reaction products
IN
     Van Wijck, Julius Gerardus Theodorus; De Haan, Andre Banier; Biermans,
    Andreas Johannes; Hardeveld, Van Rudolf
PA
    DSM N.V., Neth.
SO
    Eur. Pat. Appl., 9 pp.
    CODEN: EPXXDW
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    NL 1003105
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    Manufacture of high purity melamine from urea
ΤI
    Van Wijck, Julius Gerardus Theodorus; Sampers, Theodorus Josephus Anna
IN
    Maria; De Haan, Andre Banier
    DSM N.V., Neth.; Van Wijck, Julius Gerardus Theodorus; Sampers, Theodorus
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    Josephus Anna Maria; De Haan, Andre Banier
so
    PCT Int. Appl., 21 pp.
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TI
    Mechanistic study of fire-retardant action of melamine and its salts in
    nylon 6
    Levchik, S. V.; Balabanovich, A. I.; Levchik, G. F.; Costa, L.
ΑŲ
    Research Institute for Physical Chemical Problems of Byelorussian
    University, Minsk, Belarus
SO
    Recent Adv. Flame Retard. Polym. Mater. (1997), Volume Date 1996, 7,
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    Effect of melamine and its salts on combustion and thermal decomposition
    of polyamide 6
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    Research Institute for Physical Chemical Problems of Byelorussian
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    University, Minsk, 220080, Belarus
    Fire Mater. (1997), 21(2), 75-83
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    CODEN: FMATDV; ISSN: 0308-0501
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    Elasticized melamine resins
    Utecht, Jens; Niessner, Manfred; Kirchgaessner, Uwe; Wittmann, Otto;
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    Decher, Jakob; Jaeckh, Christof; Meier, Anton
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    BASF A.-G., Germany
    Eur. Pat. Appl., 11 pp.
SO
    CODEN: EPXXDW
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PRAI DE 1995-19535255 19950922
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    Manufacture of film-forming, aqueous, metal phosphate-amine complexes and
    their use for fire-resistant materials
    von Bonin, Wulf; Jabs, Gert; Kirchmeyer, Stephan; Raffel, Bolko; Wussow,
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    Klaus
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    Bayer A.-G., Germany
    Ger. Offen., 15 pp.
SO
    CODEN: GWXXBX
DT
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     property under ultrahigh-frequency regions
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     Koseki, Takayoshi; Yamakawa, Seishiro; Yamamoto, Hiroshi; Hirata, Isao
     Matsushita Electric Works Ltd, Japan
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ΤI
     Manufacture of guanidine and melamine phosphate
PA
     Chemie Linz (Deutschland) Gmbh I.L., Germany
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     Ger. Offen., 3 pp.
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     Process for producing high-purity melamine
TI
     Canzi, Lorenzo; Canzi, Aldo; Coufal, Gerhard; Giacomuzzo, Silvano;
IN
     Virardi, Mario; Muellner, Martin
PA
    Agrolinz Melamin Gmbh, Austria
     PCT Int. Appl., 20 pp.
SO
     CODEN: PIXXD2
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    NO 9702869
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DN
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ΤI
    Melamine of improved purity produced by high-pressure, non-catalytic
    process
IN
    Best, David; Gupta, Amit
    Melamine Chemicals, Inc., USA
PA
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    U.S., 6 pp.
    CODEN: USXXAM
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PRAI US 1995-478088 19950607
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    acid-ammonia mixture.
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    Chemie Linz GmbH, Austria
    Eur. Pat. Appl., 6 pp.
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PRAI AT 1994-1454
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    Laboratorium Anorganische Chemie, Universitaet Bayreuth, Bayreuth,
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    D-95440, Germany
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    Preparation of highly pure melamine
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IN
    Turunen, Ilkka; Oinas, Pekka
    Kemira Oy, Finland
PΑ
    PCT Int. Appl., 22 pp.
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    CODEN: PIXXD2
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    Cement dispersant containing amino acid-containing copolycondensates for
    preventing slump loss
    Yamato, Fujio; Fujita, Shuichi; Yadokoro, Yoshiaki; Sato, Haruyuki
IN
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ΡI
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AN
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DN
    Method of obtaining 2,4,6-triamino-1,3,5-triazine hydroperoxide
ΤI
    Klopotek, Alojzy; Jasion, Stefania; Osinska, Longina; Klopotek, Beata B.
IN
     Instytut Chemii Przemyslowej, Pol.
PA
     Pol., 6 pp.
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    PL 160869
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    ANSWER 47 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
    1995:383028 CAPLUS
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    Fluidized-bed reactor process for manufacturing melamine from urea at
ΤI
     elevated pressures
IN
    Lee, Jing M.
PA
    USA
SO
    U.S., 6 pp.
    CODEN: USXXAM
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L4
AN
    1994:582509 CAPLUS
DN
     121:182509
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    High-pressure noncatalytic melamine production reactor
IN
    Bizzotto, Wladimiro
PA
     Italy
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    Eur. Pat. Appl., 8 pp.
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CODEN: EPXXDW
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PRAI IT 1993-VI25
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    ANSWER 49 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
    1994:301265 CAPLUS
ΑN
DN
    120:301265
TI
    Oleoresinous-synthetic varnish preparation
IN
    Berdnikov, Mikhail P.
PA
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SO
    U.S.S.R.
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L4
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    1993:561607 CAPLUS
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DN
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    Preparation of melamine cyanurate-containing phosphate mixtures with
    ammonium, melamine, and alkaline earth metal cations
IN
    Feldmann, Walter; Schmidt, Cordelia
PΑ
    Stickstoffwerke AG Wittenberg-Piesteritz, Germany
    Ger. Offen., 6 pp.
SO
    CODEN: GWXXBX
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    ANSWER 51 OF 109 CAPLUS COPYRIGHT 2001 ACS
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DN
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    Biocidal scale inhibitors for cooling water systems
    Klopotek, Alojzy; Klopotek, Beata B.; Wlasiuk, Danuta; Haman, Waldemar;
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    Brambor, Andrzej; Murawski, Roman; Marcisiak, Jan
PΑ
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                                     PL 1987-268071 19871005
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    ANSWER 52 OF 109 CAPLUS COPYRIGHT 2001 ACS
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    1993:84038 CAPLUS
DN
    118:84038
    Propellants for gas generators, their manufacture, and safety bag systems
TI
    containing the propellants
IN
    Redecker, Klaus; Weuter, Waldemar
    Dynamit Nobel A.-G., Germany
PΑ
SO
    Eur. Pat. Appl., 16 pp.
    CODEN: EPXXDW
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    German
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    ANSWER 53 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
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DN
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TI
    Fire-retardant compositions for polyurethane foam manufacture
    Duber, Ernst Otto; Muller, Louis
IN
PA
    Imperial Chemical Industries PLC, UK
SO
    Eur. Pat. Appl., 11 pp.
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    1991:518062 CAPLUS
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    115:118062
    Removal of especially arsenic, phosphorus, sulfur, silicon, and
ΤI
molybdenum
    impurities from tungsten ores
    Grunt, Miloslav; Kodytek, Vilem
IN
PA
    Czech.
    Czech., 3 pp.
SO
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L4
AN
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    Improvement of the impact resistance of polyoxymethylene polymers
IN
    Niino, Masahiko
    Asahi Chemical Industry Co., Ltd., Japan
PA
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    Jpn. Kokai Tokkyo Koho, 15 pp.
    CODEN: JKXXAF
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    ANSWER 56 OF 109 CAPLUS COPYRIGHT 2001 ACS
L4
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    1991:410792 CAPLUS
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    115:10792
    Process for the production of aqueous solutions suitable for finishing
TI
    cellulose-containing textile materials
IN
    Bereck, Attila; Flory, Klaus; Kummer, Matthias
PA
    BASF A.-G., Fed. Rep. Ger.
    Eur. Pat. Appl., 9 pp.
SO
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    1991:251118 CAPLUS
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    Recovery of tungsten from diluted solutions
ΤI
    Kodytek, Vilem; Grunt, Miloslav
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PA
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SO
     Czech., 5 pp.
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TI
    Preparation of blocked triazine derivative polyisocyanates for use in
    coatings
    Halpaap, Reinhard; Duenwald, Wilhelm; Casselmann, Holger; Schlegel, Hans
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    Bayer A.-G., Fed. Rep. Ger.
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PRAI DE 1989-3932168 19890927
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L4
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    1991:28894 CAPLUS
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TI
    Coating of glass with transparent electric conductive film
IN
    Watanabe, Kane; Aoyama, Junichi
    Miura Printing Co., Ltd., Japan
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L4
    1990:554509 CAPLUS
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DN
    Application of a painted film to a three-dimensional object
TI
IN
    Godwin, Berner; Misev, Ljubomir
    Ciba-Geigy A.-G., Switz.
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    Eur. Pat. Appl., 6 pp.
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L4
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    Impregnation of synthetic and natural fabrics to impart fire resistance
ΤI
    and antistatic properties
IN
    Cichomski, Stanislaw; Lisiewska, Zofia; Zyska, Bronislaw; Kulawski,
Jerzy;
    Wachowicz, Jan; Stencel, Aldona
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    Yokoyama, Hiroaki; Okami, MItsuhiro
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    Zenken K. K., Japan
    Jpn. Kokai Tokkyo Koho, 4 pp.
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    1989:137107 CAPLUS
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    Thermoreactive epoxy resin powder compositions
ΤI
    Jelinek, Karel; Stary, Stanislav; Cerny, Jaroslav; Hajkova, Bohuslava;
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    Sima, Milan
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TΙ
    Admixtures for cement to improve strength
IN
    Goto, Tokio; Yamaguchi, Koichi
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    Dainippon Ink and Chemicals, Inc., Japan
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    Jpn. Kokai Tokkyo Koho, 6 pp.
    CODEN: JKXXAF
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ΑN
    1987:460030 CAPLUS
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    Utilization of melamine waste effluent
IN
    Lahalih, Shawqui; Absi-Halabi, Ma Mun
    Kuwait Institute for Scientific Research, Kuwait
PA
SO
    U.S., 9 pp.
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    Process for making solid urea-formaldehyde resins
ΤI
    Taylor, David
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    BIP Chemicals Ltd., UK
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    Eur. Pat. Appl., 14 pp.
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    CODEN: EPXXDW
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ΤI
    Flame retardants for polyolefins
    Ogawa, Yoshikatsu; Hisada, Haruhiko; Kasahara, Takeshi; Kizaki, Fumihiko;
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    Yosha, Masahide; Yoshiya, Masahide
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    Marubishi Oil Chemical Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 5 pp.
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    Microencapsulation process, multi-walled microcapsules, and transfer
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PRAI US 1980-173303 19800729
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    1986:426523 CAPLUS
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    Corrosion inhibition of heat exchangers in combustion apparatus for water
    and room heating
IN
    Fukuda, Yu; Kaneko, Yasunori
    Matsushita Electric Industrial Co., Ltd., Japan
PA
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    1986:206729 CAPLUS
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    Oxalyl fluoride
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    Nishimura, Masakatsu; Hirai, Yasuhiko
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    Tokuyama Soda Co., Ltd., Japan
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ΑN
    1986:130816 CAPLUS
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    Impregnating mixtures based on phenolic resins
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    Goetzky, Peter; Doering, Dieter; Fielitz, Ilse; Grubits, Reinhard;
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IN	Lunzer, Friedrich; Garber, Alfred									
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    Ohata, Yoichi; Ono, Takami
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    Nissan Chemical Industries, Ltd., Japan
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    Schmidt, Alfred; Wegleitner, Karlheinz; Hatzle, Josef H.; Sylpra, Rudolf;
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    Lentia G.m.b.H. Chem. u. Pharm. Erzeugnisse-Industriebedarf
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    Kokubo, Ryo; Yokomichi, Koji; Takakuwa, Yasuo; Nagakura, Mizuhiko;
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    Maruyama, Isao; Shiroishi, Akihiro
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    Melamine by urea pyrolysis
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    Vialaron, Andre
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    Fr., 8 pp.
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    Pure melamine
    Weinrotter, Ferdinand; Mueller, Walter; Schmidt, Alfred; Boehler, Walter
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PA Allied Chemical Corp.

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TI Preparation of melamine from urea at atmospheric pressure

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SO Chem.-Ing.-Tech. (1966), 38(11), 1140-4 CODEN: CITEAH

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LA German

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IN Sinn, Richard; Vogel, Ludwig; Hamprecht, Guenther; Schwarzmann, Matthias; Fromm, Hermann D.

PA Badische Anilin- und Soda-Fabrik A.-G.

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